

Synthesis and X-Ray Crystal Structural Analysis of a Fully Unsaturated Tetraphosphole Macrocycle

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The reaction of the 5,5'-bis(formyl)-2,2'-biphosphole **1** with the *o*-xylylidene bis-ylide **2** affords the 24-membered tetraphosphole macrocycle **3** in 60% yield; the X-ray crystal structure of **3** shows an all-*trans* disposition of the four lone pairs and a diameter of the central cavity of ca. 6 Å.

Pyrroles form the basis of a wide range of macrocycles, which include the porphyrins, porphycenes and expanded porphyrins.¹ Their phosphorus analogues are unknown. With a correct tailoring of the macrocycle, taking into account the larger size of the phosphole ring than its nitrogen counterpart,² this kind of rigid tetradentate crown should certainly have a rich coordination chemistry, particularly towards soft, catalytically active, heavy transition metal centres. With such a goal in mind, we describe here the synthesis and X-ray crystal structure analysis of the first α,α' -connected tetraphosphole macrocycle. Our

starting point is the 5,5'-biscarboxaldehyde **1**.³ Upon reaction with the bis-ylide **2** at low temperature, it affords the expected tetraphosphole macrocycle **3** in a surprisingly high yield (60%).[†]

The X-ray crystal structure analysis of **3**‡ shows a distorted 24-membered macrocycle with an all-*trans*-disposition of the four phosphorus-phenyl substituents which provide some steric protection to the central cavity. This hole is rather large as shown by the diagonal P...P distances ca. 6.1 Å. Otherwise, the four phosphole rings display quite normal bond distances and angles.

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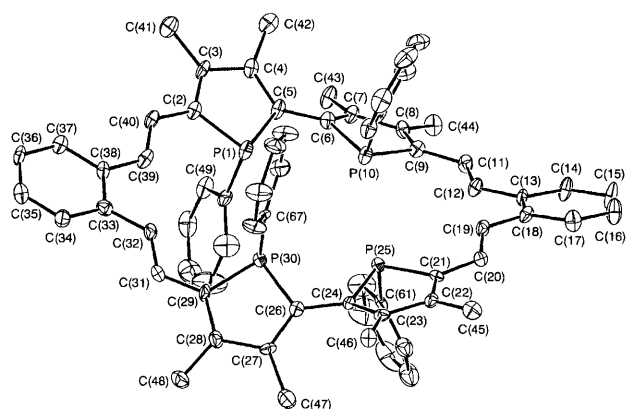
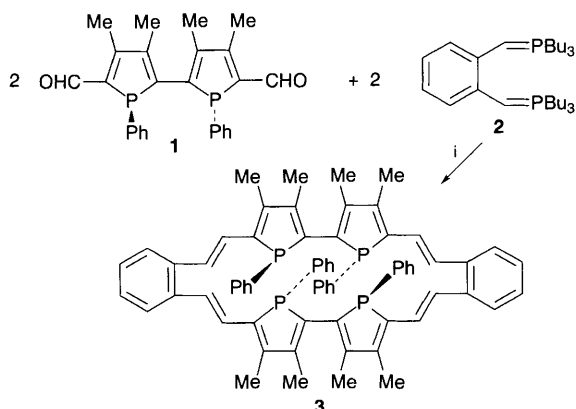


Fig. 1 Molecular structure of $C_{68}H_{60}P_4$ **3**, as determined by a single X-ray diffraction study. Selected bond lengths, distances (Å) and angles ($^\circ$): P(1)–C(2) 1.807(5), P(1)–C(5) 1.822(5), P(1)–C(49) 1.830(6), C(2)–C(3) 1.356(7), C(2)–C(40) 1.438(7), C(3)–C(4) 1.469(6), C(4)–C(5) 1.364(6), C(5)–C(6) 1.450(6), C(33)–C(38) 1.428(7), C(38)–C(39) 1.473(7), C(39)–C(40) 1.336(7), P₁...P₁₀ 3.307, P₁...P₃₀ 5.202, P₁₀...P₂₅ 5.060, P₂₅...P₃₀ 3.302, P₁...P₂₅ 6.056, P₁₀...P₃₀ 6.148, P(1)C(2)–C(5)/P(10)C(6)–C(9) 35.90 \pm 0.24, P(25)C(21)–C(24)/P(30)C(26)–C(29) 38.66 \pm 0.14, P(1)C(2)–C(5)/C(33)–C(38) 14.36 \pm 0.52, P(10)C(6)–C(9)/C(13)–C(18) 16.81 \pm 0.43, P(25)C(21)–C(24)/C(13)–C(18) 15.39 \pm 0.43, P(30)C(26)–C(29)/C(33)–C(38) 28.46 \pm 0.23.



Scheme 1 Reagents and conditions: i, THF, addition of **1** to **2** at -50 $^\circ\text{C}$, then back to $+25$ $^\circ\text{C}$

Footnotes

[†] The bis-ylide **2** is obtained by reaction of the corresponding diphosphonium dibromide with *n*-butyllithium at -50 $^\circ\text{C}$ in THF until complete dissolution: δ ^{31}P **2** +7.4. See ref. 4 for synthesis of the Ph_3P analogue.

The crude macrocycle **3** is purified by chromatography on neutral alumina with hexane:diethyl ether 85:15 as the eluent. Deep-red crystals; ^{31}P NMR (CH_2Cl_2): δ +15.9; MS(EI): m/z 1000 (M^+ , 100%), 500 (100%).

[‡] *Crystal data* for **3**: Crystals of $C_{68}H_{60}P_4 \cdot 1.5\text{CH}_2\text{Cl}_2$ were grown from a CH_2Cl_2 -MeOH solution of the compound. Data were collected at -150 ± 0.5 $^\circ\text{C}$ on an Enraf-Nonius CAD4 diffractometer using $\text{Cu-K}\alpha$ radiation (λ 1.54184 Å) and a graphite monochromator. The crystal structure was solved and refined using the Enraf-Nonius MOLEN package. The compound crystallises in space group $P\bar{1}$ (#2), $a = 14.132(1)$, $b = 16.884(2)$, $c = 25.994(3)$ Å, $\alpha = 76.02(1)$, $\beta = 79.03(1)$, $\gamma = 85.36(1)^\circ$; $V = 5904.58$ (1.3) Å³; $Z = 4$; $D_c = 1.269$ g cm^{-3} ; $\mu = 27.8$ cm^{-1} ; $F(000) = 2364$. A total of 10077 unique reflections were recorded in the range $3^\circ \leq 2\theta \leq 100.0^\circ$ of which 2862 were considered as unobserved [$F^2 < 3.0\sigma(F^2)$], leaving 7215 for solution and refinement. Direct methods yielded a solution for all atoms. The hydrogen atoms were included as fixed contributions in the final stages of least-squares refinement while using anisotropic temperature factors for all other atoms except one CH_2Cl_2 carbon atom. The latter could not be refined anisotropically as a result of a non-resolvable disorder. Aside from minor conformational differences, the two independent molecules present in the asymmetric unit are identical. A non-Poisson weighting scheme was applied with a p factor equal to 0.08. The final agreement factors were $R = 0.048$, $R_w = 0.070$, $\text{GOF} = 1.54$. Atomic coordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Information for Authors, Issue No. 1, 1995.

References

- For a short but recent overview of the field, see: J. L. Sessler, S. J. Waghorn, Y. Hiseada and V. Lynch, *Chem. Eur. J.*, 1995, **1**, 56.
- For comparison: pyrrole: intracyclic $< \text{CNC}$ ca. 109° , intracyclic N–C ca. 1.37 Å, N covalent radius 0.75 Å; phosphole: intracyclic $< \text{CPC}$ ca. 90° , intracyclic P–C ca. 1.78 Å, P covalent radius 1.10 Å; see: P. Coggon and A. T. McPhail, *J. Chem. Soc., Dalton Trans.*, 1973, 1888.
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